

# Material Safety Data Sheet

HAZARD WARNINGS	RISK PHRASES	PROTECTIVE CLOTHING
	Flammable material; avoid heat and sources of ignition. Harmful compound, minimize exposure. Irritating to skin, eyes, and the respiratory system. <b>POSSIBLE CARCINOGEN. MINIMIZE EXPOSURE.</b> Hygroscopic -- keep container tightly sealed. Light sensitive. Store under nitrogen.	

## Section I. Chemical Product and Company Identification

Chemical Name	<b>Ninhydrin</b> (contains Acetic Acid) Ethanol Solution [for TLC Stain]		
Catalog Number	N0719	Supplier	TCI America 9211 N. Harborsgate St. Portland OR 1-800-423-8616
Synonym	Not available.		
Chemical Formula	C <sub>9</sub> H <sub>6</sub> O <sub>4</sub>		
CAS Number	485-47-2 (Ninhydrin) 64-17-5 (Ethanol)	In case of Emergency Call	<b>Chemtrec®</b> <b>(800) 424-9300 (U.S.)</b> <b>(703) 527-3887 (International)</b>

## Section II. Composition and Information on Ingredients

Chemical Name	CAS Number	Percent (%)	TLV/PEL	Toxicology Data
Ninhydrin (contains Acetic Acid) Ethanol Solution [for TLC Stain]	485-47-2 (Ninhydrin) 64-17-5 (Ethanol)	Not available.	This chemical is classified as a possible carcinogen. There is no acceptable exposure limit for a carcinogen.	(Ninhydrin) Mouse LD <sub>50</sub> (intraperitoneal) 78 mg/kg (Ethanol) Rat LD <sub>50</sub> (oral) 7060 mg/kg Mouse LD <sub>50</sub> (oral) 3450 mg/kg Rat LD <sub>50</sub> (inhalation) 20000 ppm/10H

## Section III. Hazards Identification

Acute Health Effects	Harmful if ingested or inhaled. Minimize exposure to this material. Severe overexposure can result in injury or death. Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering. Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.
Chronic Health Effects	<b>CARCINOGENIC EFFECTS</b> : Not available. <b>MUTAGENIC EFFECTS</b> : Not available. <b>TERATOGENIC EFFECTS</b> : Tumorigenic effects. (Ethanol) Mouse TD Oral 400 gm/kg for 57 weeks intermittent <b>TOXIC EFFECTS:</b> Tumorigenic - Equivocal tumorigenic agent by RTECS criteria Gastrointestinal - Tumors Mouse TDLo Oral 320 mg/kg for 50 weeks intermittent <b>TOXIC EFFECTS:</b> Tumorigenic - Equivocal tumorigenic agent by RTECS criteria Liver - Tumors Blood - Lymphomas including Hodgkin's disease Mouse TDLo Rectal 120 gm/kg for 18 weeks intermittent <b>TOXIC EFFECTS:</b> Tumorigenic - Equivocal tumorigenic agent by RTECS criteria Gastrointestinal - Tumors Liver - Tumors <b>DEVELOPMENTAL TOXICITY:</b> Reproductive effects. (Ethanol) Rat TDLo Intraperitoneal 600 mg/kg female 8-15 days of pregnancy <b>TOXIC EFFECTS:</b> Effects on Fertility - Post-implantation mortality Effects on Embryo or Fetus - Extra embryonic structures Effects on Embryo or Fetus - Fetotoxicity Rat TDLo Oral 135 gm/kg female 1 day of pregnancy to 7 days after birth <b>TOXIC EFFECTS:</b> Effects on Newborn - Behavioral Effects on Newborn - Physical Rat TDLo Oral 147 mg/kg female 1-21 days of pregnancy <b>TOXIC EFFECTS:</b> Specific Developmental Abnormalities - Endocrine system Effects on Newborn - Delayed effects Repeated or prolonged exposure to this compound is not known to aggravate existing medical conditions.

(contains Acetic Acid) Ethanol Solution  
[for TLC Stain]**Section IV. First Aid Measures**

Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention.
Skin Contact	In case of contact, immediately flush skin with plenty of water. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.
Inhalation	If the victim is not breathing, perform mouth-to-mouth resuscitation. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, oxygen can be administered. Seek medical attention if respiration problems do not improve.
Ingestion	INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive.

**Section V. Fire and Explosion Data**

Flammability	Flammable.	Auto-Ignition	363 °C (685.4 °F) (Ethanol)
Flash Points	14 °C (57.2 °F). (Ethanol)	Flammable Limits	LOWER: 3.3% UPPER: 19% (Ethanol)
Combustion Products	These products are toxic carbon oxides (CO, CO <sub>2</sub> ).		
Fire Hazards	Not available.		
Explosion Hazards	Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.		
Fire Fighting Media and Instructions	Flammable liquid. SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use alcohol foam, water spray or fog. Consult with local fire authorities before attempting large scale fire-fighting operations.		

**Section VI. Accidental Release Measures**

Spill Cleanup Instructions	Flammable material. Harmful material. Irritating material. Possibly carcinogenic material. Hygroscopic material. Light sensitive material. Keep away from heat. Mechanical exhaust required. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. DO NOT touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Consult federal, state, and/or local authorities for assistance on disposal.
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**Section VII. Handling and Storage**

Handling and Storage Information	FLAMMABLE. HARMFUL. IRRITANT. POSSIBLE CARCINOGEN. HYGROSCOPIC. LIGHT SENSITIVE. STORE UNDER NITROGEN. Keep away from heat. Mechanical exhaust required. Avoid excessive heat and light. Do not breathe gas/fumes/ vapor/spray. Always store away from incompatible compounds such as oxidizing agents.
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**Section VIII. Exposure Controls/Personal Protection**

Engineering Controls	Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash station and safety shower is proximal to the work-station location.
Personal Protection	Splash goggles. Lab coat. Vapor respirator. Boots. Gloves. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product. Be sure to use a MSHA/NIOSH approved respirator or equivalent.



Exposure Limits	This chemical is classified as a possible carcinogen. There is no acceptable exposure limit for a carcinogen.
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**Section IX. Physical and Chemical Properties**

Physical state @ 20°C	Liquid. (Clear, light yellow.)	Solubility	Ninhydrin is soluble in water, alcohols.
Specific Gravity	0.79 (water=1) (Ethanol)		
Molecular Weight	C <sub>9</sub> H <sub>6</sub> O <sub>4</sub> = 178.14 (Ninhydrin) C <sub>2</sub> H <sub>6</sub> O = 46.07 (Ethanol)	Partition Coefficient	LOG P <sub>ow</sub> : -0.32 (Ethanol)
Boiling Point	79 °C (174.2 °F) (Ethanol)	Vapor Pressure	5.8 kPa (@ 20 °C) (Ethanol)
Melting Point	250 °C (482 °F) (Ninhydrin)	Vapor Density	1.6 (Air = 1) (Ethanol)
Refractive Index	Not available.	Volatility	Not available.
Critical Temperature	Not available.	Odor	Characteristic.
Viscosity	Not available.	Taste	Not available.

(contains Acetic Acid) Ethanol Solution  
[for TLC Stain]**Section X. Stability and Reactivity Data**

Stability	This material is stable if stored under proper conditions. (See Section VII for instructions)
Conditions of Instability	Avoid excessive heat and light. Hygroscopic; keep container tightly closed. Store under nitrogen. Sensitive to light.
Incompatibilities	Reactive with oxidizing agents, peroxides, ammonia, alkali metals.

**Section XI. Toxicological Information**

RTECS Number	NK5425000 (Ninhydrin) KQ6300000 (Ethanol)
Routes of Exposure	Eye Contact. Ingestion. Inhalation.
Toxicity Data	(Ninhydrin) Mouse LD <sub>50</sub> (intraperitoneal) 78 mg/kg (Ethanol) Rat LD <sub>50</sub> (oral) 7060 mg/kg Mouse LD <sub>50</sub> (oral) 3450 mg/kg Rat LD <sub>50</sub> (inhalation) 20000 ppm/10H
Chronic Toxic Effects	<b>CARCINOGENIC EFFECTS</b> : Not available. <b>MUTAGENIC EFFECTS</b> : Not available. <b>TERATOGENIC EFFECTS</b> : Tumorigenic effects. (Ethanol) Mouse TD Oral 400 gm/kg for 57 weeks intermittent <b>TOXIC EFFECTS:</b> Tumorigenic - Equivocal tumorigenic agent by RTECS criteria Gastrointestinal - Tumors Mouse TDLo Oral 320 mg/kg for 50 weeks intermittent <b>TOXIC EFFECTS:</b> Tumorigenic - Equivocal tumorigenic agent by RTECS criteria Liver - Tumors Blood - Lymphomas including Hodgkin's disease Mouse TDLo Rectal 120 gm/kg for 18 weeks intermittent <b>TOXIC EFFECTS:</b> Tumorigenic - Equivocal tumorigenic agent by RTECS criteria Gastrointestinal - Tumors Liver - Tumors <b>DEVELOPMENTAL TOXICITY:</b> Reproductive effects. (Ethanol) Rat TDLo Intraperitoneal 600 mg/kg female 8-15 days of pregnancy <b>TOXIC EFFECTS:</b> Effects on Fertility - Post-implantation mortality Effects on Embryo or Fetus - Extra embryonic structures Effects on Embryo or Fetus - Fetotoxicity Rat TDLo Oral 135 gm/kg female 1 day of pregnancy to 7 days after birth <b>TOXIC EFFECTS:</b> Effects on Newborn - Behavioral Effects on Newborn - Physical Rat TDLo Oral 147 mg/kg female 1-21 days of pregnancy <b>TOXIC EFFECTS:</b> Specific Developmental Abnormalities - Endocrine system Effects on Newborn - Delayed effects Repeated or prolonged exposure to this compound is not known to aggravate existing medical conditions.
Acute Toxic Effects	Harmful if ingested or inhaled. Minimize exposure to this material. Severe overexposure can result in injury or death. Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering. Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

**Section XII. Ecological Information**

Ecotoxicity	Not available.
Environmental Fate	Ethanol's production and use in alcoholic beverages, as a solvent, fuel additive, in the manufacture of denatured alcohol, pharmaceuticals (rubbing compounds, tonics, lotions, colognes), in perfumery, and organic synthesis may result in its release to the environment through various waste streams; its use as a fungicide and plant regulator will result in its direct release to the environment. Ethanol has been identified as a natural emission product from various plants, fermentation product and as a biological decomposition product of wastes and sewage. If released to the atmosphere, an extrapolated vapor pressure of 59.3 mm Hg at 25 deg C indicates that ethanol will exist solely in the vapor phase. Vapor phase ethanol is degraded in the atmosphere by reaction with photochemically-produced hydroxyl radicals; the half-life for this reaction in air is estimated to be 5 days. If released to soil, ethanol is expected to have very high mobility based upon an estimated Koc of 1. Volatilization from moist soil surfaces is expected to be an important fate process based upon a Henry's Law constant of 5X10 <sup>-6</sup> atm-cu m/mole. Ethanol may also volatilize from dry soils based upon its vapor pressure. Biodegradation is expected to occur rapidly in the environment based on numerous screening tests using different types of inocula and incubation periods. Ethanol was degraded with half-lives on the order of a few days using microcosms constructed with a low organic sandy soil and groundwater, indicating it is unlikely to be persistent in the environment. If released into water, ethanol is not expected to adsorb to suspended solids and sediment based upon the estimated Koc. Volatilization from water surfaces is expected to be an important fate process based upon this compound's Henry's Law constant. Estimated volatilization half-lives for a model river and model lake are 3 and 39 days, respectively. An estimated BCF of 3 suggests the potential for bioconcentration in aquatic organisms is low. Hydrolysis of ethanol and photolysis in sunlit surface waters are not expected since ethanol lacks functional groups that are susceptible to hydrolysis or photolysis under environmental conditions. Occupational exposure to ethanol may occur through inhalation and dermal contact with this compound at workplaces where ethanol is produced or used. The general population is directly exposed to ethanol through the consumption of alcoholic beverages and other products that contain ethanol. Monitoring data also indicate that the general population may be exposed to ethanol via inhalation of ambient air.

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Emergency phone number (800) 424-9300

(contains Acetic Acid) Ethanol Solution  
[for TLC Stain]**Section XIII. Disposal Considerations**

Waste Disposal	Recycle to process, if possible. Consult your local regional authorities. You may be able to dissolve or mix material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber system. Observe all federal, state and local regulations when disposing of the substance.
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**Section XIV. Transport Information**

DOT Classification	DOT CLASS 3: Flammable liquid.
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PIN Number	UN1170
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Proper Shipping Name	Ethanol solution
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Packing Group (PG)	II
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DOT Pictograms	
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**Section XV. Other Regulatory Information and Pictograms**

TSCA Chemical Inventory (EPA)	This compound is <b>ON</b> the EPA Toxic Substances Control Act (TSCA) inventory list.
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WHMIS Classification (Canada)	CLASS B-2: Flammable liquid with a flash point lower than 37.8°C (100°F). On DSL.
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EINECS Number (EEC)	207-618-1 (Ninhydrin) 200-578-6 (Ethanol)
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EEC Risk Statements	R11- Highly flammable. R18- In use, may form flammable/explosive vapor-air mixture. R20/21/22- Harmful by inhalation, in contact with skin and if swallowed. R36/37/38- Irritating to eyes, respiratory system and skin. R45- May cause cancer.
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Japanese Regulatory Data	ENCS No. 4-584 (Ninhydrin) ENCS No. 2-202 (Ethanol)
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**Section XVI. Other Information**

**Version 1.0**  
**Validated on 1/14/2009.**  
**Printed 1/14/2009.**

**Notice to Reader**

TCI laboratory chemicals are for research purposes only and are NOT intended for use as drugs, food additives, households, or pesticides. The information herein is believed to be correct, but does not claim to be all inclusive and should be used only as a guide. Neither the above named supplier nor any of its subsidiaries assumes any liability whatsoever for the accuracy or completeness of the information contained herein. Final determination of suitability of any material is the sole responsibility of the user. All chemical reagents must be handled with the recognition that their chemical, physiological, toxicological, and hazardous properties have not been fully investigated or determined. All chemical reagents should be handled only by individuals who are familiar with their potential hazards and who have been fully trained in proper safety, laboratory, and chemical handling procedures. Although certain hazards are described herein, we can not guarantee that these are the only hazards which exist. Our MSDS sheets are based only on data available at the time of shipping and are subject to change without notice as new information is obtained. Avoid long storage periods since the product is subject to degradation with age and may become more dangerous or hazardous. It is the responsibility of the user to request updated MSDS sheets for products that are stored for extended periods. Disposal of unused product must be undertaken by qualified personnel who are knowledgeable in all applicable regulations and follow all pertinent safety precautions including the use of appropriate protective equipment (e.g. protective goggles, protective clothing, breathing equipment, facial mask, fume hood). For proper handling and disposal, always comply with federal, state, and local regulations.