



Material Safety Data Sheet

HAZARD WARNINGS





Flammable material; avoid heat and sources of ignition. Harmful compound, minimize exposure.

RISK PHRASES

Irritating to skin, eyes, and the respiratory system.

Lachrymator. Environmental hazard.

Refrigerate.





PROTECTIVE CLOTHING



Section I. Chemical Product and Company Identification				
Chemical Name	Methyl Methacrylate, (stabilized with 6-tert-Butyl-2,4-xylenol)			
Catalog Number	M0087	Supplier	TCI America 9211 N. Harborgate St.	
Synonym	Methyl 2-methyl-2-propenoate		Portland OR 1-800-423-8616	
Chemical Formula	$C_5H_8O_2$			
CAS Number	80-62-6	In case of Emergency Call	Chemtrec® (800) 424-9300 (U.S.) (703) 527-3887 (International)	

Section II. Composition and Information on Ingredients					
Chemical Name	CAS Number	Percent (%)	TLV/PEL	Toxicology Data	
Methyl Methacrylate, (stabilized with 6-tert-butyl-2,4-xylenol)	80-62-6	Min. 99.8%(GC)		Rat LD_{50} (oral) 7872 mg/kg Rabbit LD_{50} (dermal) >5 g/kg Rat LC_{50} (inhalation) 78000 mg/m ³	

Hazards Identification Section III.

Acute Health Effects

Harmful if ingested or inhaled. Minimize exposure to this material. Severe overexposure can result in injury or death. Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening. or, occasionally, blistering.

Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

Chronic Health Effects

CARCINOGENIC EFFECTS: Not available.

MUTAGENIC EFFECTS: Not available.
TERATOGENIC EFFECTS: TUMORIGENIC EFFECTS

Rat TDLo Implant; 1620 mg/kg

TOXIC EFFECTS

Tumorigenic - Equivocal tumorigenic agent by RTECS criteria

Tumorigenic - Tumors at site of application

DEVELOPMENTAL TOXICITY: REPRODUCTIVE EFFECTS Woman TCLo Inhalation; 10 mg/m3; female 9 years of pregnancy

TOXIC EFFECTS

Maternal Effects - Other effects

Effects on Embryo or Fetus - Extra embryonic structures (e.g., placenta, umbilical cord)

Effects on Newborn - Delayed effects

Rat TCLo Inhalation; 54 mg/m3/24 hours; female 8 weeks prior to mating

TOXIC EFFECTS

Maternal Effects - Menstrual cycle changes or disorders

Mouse TCLo Inhalation; 116 ppm/6 hours; female 4 to 13 days of pregnancy

Effects on Fertility - Post implantation mortality (e.g., dead and or resorbed implants per total number of implants)

Effects on Embryo or Fetus - Fetotoxicity (except death, e.g., stunted fetus)

Specific Developmental Abnormalities - Musculoskeletal system

Repeated or prolonged exposure to this compound is not known to aggravate existing medical conditions.

Section IV.	First Aid Measures
Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention.
Skin Contact	In case of contact, immediately flush skin with plenty of water. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.
Inhalation	If the victim is not breathing, perform mouth-to-mouth resuscitation. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, oxygen can be administered. Seek medical attention if respiration problems do not improve.

INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive.

Ingestion

M0087	Methyl Methacrylate, (stabilized with 6-tert-Butyl-2,4-xylenol) Page 2						
Section V. Fire and Explosion Data							
Flammability	Flammable.	Auto-Ignition	421 °C (789.8 °F)				
Flash Points	9℃ (48.2℃).	Flammable Limits	LOWER: 1.7% UPPER: 12.5%				
Combustion Products	These products are toxic carbon oxides	(CO, CO ₂).					
Fire Hazards	Not available.						
Explosion Hazards		Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.					
Fire Fighting Media and Instructions	Flammable liquid. SMALL FIRE: Use DRY chemical powder LARGE FIRE: Use alcohol foam, water consult with local fire authorities before	spray or fog.	operations.				
Section VI.	Accidental Release Measure	es					
Spill Cleanup Instructions	Flammable Material. Harmful Material. Irritating Material. Lachrymatory. Environmental hazard. Keep away from heat. Mechanical exhaust required. Stop leak if without risk. Absorb with DRY earth, sand or other non-combustible material. DO NOT touch spilled material. Prevent entry into sewers, basements or confined areas; dike if needed. Consult federal, state, and/or local authorities for assistance on disposal.						
Section VII.	Handling and Storage						
Handling and Storage Information	from heat. Mechanical exhaust required	FLAMMABLE. HARMFUL. IRRITANT. LACHRYMATORY. ENVIRONMENTAL HAZARD. REFRIGERATE. Keep away from heat. Mechanical exhaust required. Avoid excessive heat and light. Do not breathe gas/fumes/ vapor/spray. Always store away from incompatible compounds such as oxidizing agents, reducing agents, alkalis (bases), moisture.					
Section VIII.	Exposure Controls/Personal Protection						
Engineering Controls		Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash station and safety shower is proximal to the work-station location.					
Personal Protection		Splash goggles. Lab coat. Vapor respirator. Boots. Gloves. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product. Be sure to use a MSHA/NIOSH approved respirator or equivalent.					
Exposure Limits	Not available.						
Section IX.	Physical and Chemical Prop	perties					
Physical state @ 20°C	Liquid. (Clear, Colorless.)	Solubility	Soluble in alcohol, ether, acetone,				
Specific Gravity	0.943 (water=1)	_	chlorinated hydrocarbons, methyl ethyl ketone, and other organic solvents. Slightly soluble in water (1.6g/100ml @ 20°C), glycol.				
Molecular Weight	100.12	Partition Coefficient	Log P _{ow} 1.38				
Boiling Point	100℃ (212℉)	Vapor Pressure	3.9 kPa @ 20 ℃				
Melting Point	-48 °C (-54.4 °F)	Vapor Density	3.5 (Air = 1)				
Refractive Index	1.413 to 1.416	– Volatility	Not available.				
Critical Temperature	Not available.	Odor	Characteristic.				
Viscosity	0.63 pas (@ 20℃)	Taste	Not available.				
Section X. Stability and Reactivity Data							
Stability	This material is stable if stored under proper conditions. (See Section VII for instructions)						
Conditions of Instability	May polymerize on exposure to light. Avoid excessive heat and light.						
Incompatibilities	Poactive with evidizing agents, reducing agents, strong acids, strong alkalis (bases), moisture, amines, balegons						

Section XI. Toxicological Information

RTECS Number

OZ5075000

Routes of Exposure

Eye Contact. Ingestion. Inhalation.

Toxicity Data

Rat LD_{50} (oral) 7872 mg/kg Rabbit LD_{50} (dermal) >5 g/kg Rat LC_{50} (inhalation) 78000 mg/m³/4H

Chronic Toxic Effects

CARCINOGENIC EFFECTS: Not available. **MUTAGENIC EFFECTS**: Not available.

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Section XII. Ecological Information

Ecotoxicity

Not available

Environmental Fate

Methyl methacrylate's production and use in polymethacrylate resins, in medicinal adhesives, dental technology, bone cements, and as a water-repellent on concrete surfaces may result in its release to the environment through various waste streams. If released to air, a vapor pressure of 38.5 mm Hg at 25 deg C indicates methyl methacrylate will exist solely as a vapor in the ambient atmosphere. Vapor-phase methyl methacrylate will be degraded in the atmosphere by reaction with photochemically-produced hydroxyl radicals; the half-life for this reaction in air is estimated to be 7.4 hours. Because methyl methacrylate does not absorb light in the environmental UV spectrum (>290 nm), it is not expected to directly photolyze. If released to soil, methyl methacrylate is expected to have high mobility based upon a Koc of 95. Volatilization from moist soil surfaces is expected to be an important fate process based upon an estimated Henry's Law constant of 3.2X10-4 atm-cu m/mole. Methyl methacrylate may potentially volatilize from dry soil surfaces based upon its vapor pressure. If released into water, methyl methacrylate is not expected to adsorb to suspended solids and sediment in the water column based upon the estimated Koc. Screening studies support rapid biodegradation of methyl methacrylate; it reached 94% of its theoretical BOD in 2 weeks using an activated sludge inoculum. Volatilization from water surfaces is expected to be an important fate process based upon this compound's estimated Henry's Law constant. Estimated volatilization half-lives for a model river and model lake are 6 hours and 5 days, respectively. An estimated BCF of 7 suggests the potential for bioconcentration in aquatic organisms is low. Hydrolysis of methyl methacrylate may be a significant process under basic conditions based upon a hydrolytic half-life of 3.4 hours at pH 11; half-lives at pH 7, 8, and 9 were 4 years, 140 days, and 14 days respectively. Occupational exposure to methyl methacrylate may occur through inhalation and dermal contact with this compound at workplaces where methyl methacrylate is produced or used. The general population may be exposed to methyl methacrylate via ingestion of drinking water and inhalation or dermal contact with resins, dental products, or artificial nail products containing methyl methacrylate.

Section XIII. Disposal Considerations

Waste Disposal

Recycle to process, if possible. Consult your local regional authorities. You may be able to dissolve or mix material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber system. Observe all federal, state and local regulations when disposing of the substance.

Section XIV. Transport Information

DOT Classification

DOT Class 3: Flammable liquid.

PIN Number

UN1247

Proper Shipping Name

Methyl methacrylate monomer, stabilized

Packing Group (PG)

п

DOT Pictograms



Emergency phone number (800) 424-9300

Section XV. Other Regulatory Information and Pictograms TSCA Chemical Inventory This compound is ON the EPA Toxic Substances Control Act (TSCA) inventory list. (EPA) WHMIS Classification CLASS B-2: Flammable liquid with a flash point lower than 37.8 °C (100 °F). On DSL. (Canada) EINECS Number (EEC) 201-297-1 EEC Risk Statements R10- Flammable. R18- In use, may form flammable/explosive vapor-air mixture. R20/21/22- Harmful by inhalation, in contact with skin and if swallowed. R36/37/38- Irritating to eyes, respiratory system and skin. R50- Very toxic to aquatic organisms. R51- Toxic to aquatic organisms. R52- Harmful to aquatic organisms. R53- May cause long-term adverse effects in the aquatic environment. Japanese Regulatory Data ENCS No. (2)-1036

Section XVI. Other Information

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Notice to Reader

TCI laboratory chemicals are for research purposes only and are NOT intended for use as drugs, food additives, households, or pesticides. The information herein is believed to be correct, but does not claim to be all inclusive and should be used only as a guide. Neither the above named supplier nor any of its subsidiaries assumes any liability whatsoever for the accuracy or completeness of the information contained herein. Final determination of suitability of any material is the sole responsibility of the user. All chemical reagents must be handled with the recognition that their chemical, physiological, toxicological, and hazardous properties have not been fully investigated or determined. All chemical reagents should be handled only by individuals who are familiar with their potential hazards and who have been fully trained in proper safety, laboratory, and chemical handling procedures. Although certain hazards are described herein, we can not guarantee that these are the only hazards which exist. Our MSDS sheets are based only on data available at the time of shipping and are subject to change without notice as new information is obtained. Avoid long storage periods since the product is subject to degradation with age and may become more dangerous or hazardous. It is the responsibility of the user to request updated MSDS sheets for products that are stored for extended periods. Disposal of unused product must be undertaken by qualified personnel who are knowledgeable in all applicable regulations and follow all pertinent safety precautions including the use of regulations.

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