



Material Safety Data Sheet

PROTECTIVE CLOTHING HAZARD WARNINGS RISK PHRASES Harmful compound, minimize exposure. Irritating to skin, eyes, and the respiratory system. CARCINOGEN. MINIMIZE EXPOSURE.

Section I. Chemical Product and Company Identification				
Chemical Name	Dichloromethane Anhydrous(stabilized with 2-Methyl-2-butene)			
Catalog Number	D3478	Supplier	TCI America 9211 N. Harborgate St.	
Synonym	Methylene Chloride Anhydrous		Portland OR 1-800-423-8616	
Chemical Formula	CH ₂ C ₂		***************************************	
CAS Number	75-09-2	In case of Emergency	Chemtrec® (800) 424-9300 (U.S.)	
		Call	(703) 527-3887 (International)	

Section II. Composition and Information on Ingredients				
Chemical Name	CAS Number	Percent (%)	TLV/PEL	Toxicology Data
Dichloromethane Anhydrous _(stabilized with 2-Methyl-2-butene)	75-09-2	Min. 99.0%(GC)	This chemical is classified as a carcinogen. There is no acceptable exposure limit for a carcinogen.	Rabbit LD ₅₀ (oral) 2000 mg/kg

Section III.	Hazards Ide	ntification
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Harmful if ingested or inhaled. Minimize exposure to this material. Severe overexposure can result in injury or death. Acute Health Effects

Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening,

or, occasionally, blistering.

Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

CARCINOGENIC EFFECTS: Carcinogenic by RTECS criteria. Chronic Health Effects

MUTAGENIC EFFECTS: Not available.

TERATOGENIC EFFECTS: TUMORIGENIC EFFECTS Rat TCLo Inhalation; 3500 ppm/6 hours/2 years intermittent

TOXIC EFFECTS

Tumorigenic - Carcinogenic by RTECS criteria

Endocrine - Tumors

TCLo Inhalation; 102000 mg/kg/102 weeks intermittent

TOXIC EFFECTS

Tumorigenic - Carcinogenic by RTECS criteria

Skin and Appendages - Tumors

Blood - Leukemia

Mouse TCLo Inhalation; 122400 mg/kg/102 weeks intermittent

TOXIC EFFECTS

Tumorigenic - Carcinogenic by RTECS criteria

Lung, Thorax, or Respiration - Tumors

Liver - Tumors

DEVELOPMENTAL TOXICITYREPRODUCTIVE EFFECTS

Rat TCLo Inhalation; 4500 ppm/24 hours; female 1 to 17 days of pregnancy

TOXIC EFFECTS

Effects on Newborn - Behavioral

TCLo Inhalation; 1250 ppm/7 hours; female 6 to 15 days of pregnancy

TOXIC EFFECTS

Specific Developmental Abnormalities - Musculoskeletal system

Specific Developmental Abnormalities - Urogenital system

Mouse TCLo Inhalation; 1250 ppm/7 hours; female 6 to 15 days of pregnancy

TOXIC EFFECTS

Specific Developmental Abnormalities - Musculoskeletal system

Repeated exposure to an highly toxic material may produce general deterioration of health by an accumulation in one or many human organs.

Section IV. First Aid Measures

Eye Contact Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15

minutes. Get medical attention.

Skin Contact In case of contact, immediately flush skin with plenty of water for at least 15 minutes while removing contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention immediately.

If the victim is not breathing, perform mouth-to-mouth resuscitation. Loosen tight clothing such as a collar, tie, belt or Inhalation waistband. If breathing is difficult, oxygen can be administered. Seek medical attention if respiration problems do not

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D3478	Dichloromethane Anhydrous _(stabilized with 2-Methyl-2-butene) Page 2
Ingestion	INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive.
Section V.	Fire and Explosion Data

Section V. F	ire and Explosion Data		
Flammability	May be combustible at high temperature.	Auto-Ignition	556.1°C (1033°F)
Flash Points	100°C (212°F).	Flammable Limits	LOWER: 12% UPPER: 25%
Combustion Products	These products are toxic carbon oxides (CO,	CO ₂).	
Fire Hazards	Not available.		
Explosion Hazards	Risks of explosion of the product in presence of mechanical impact: Not available. Risks of explosion of the product in presence of static discharge: Not available.		
Fire Fighting Media and Instructions	SMALL FIRE: Use DRY chemical powder. LARGE FIRE: Use water spray, fog or foam. Consult with local fire authorities before attem		operations.

Section VI. Accidental Release Measures

Spill Cleanup Instructions

Harmful Material. Irritating Material. Carcinogenic Material.

Stop leak if without risk. DO NOT get water inside container. DO NOT touch spilled material. Use water spray to reduce vapors. Prevent entry into sewers, basements or confined areas; dike if needed. Eliminate all sources of ignition. Consult federal, state, and/or local authorities for assistance on disposal.

Section VII. Handling and Storage

Handling and Storage Information

HARMFUL. IRRITANT. CARCINOGEN. Keep locked up.. Keep away from heat. Mechanical exhaust required. When not in use, tightly seal the container and store in a dry, cool place. Avoid excessive heat and light. DO NOT ingest. Do not breathe gas/fumes/ vapor/spray. Wear suitable protective clothing. If ingested, seek medical advice immediately and show the container or the label. Treat symptomatically and supportively. Always store away from incompatible compounds such as oxidizing agents, metals, alkalis (bases)

Section VIII. Exposure Controls/Personal Protection

Engineering Controls

Provide exhaust ventilation or other engineering controls to keep the airborne concentrations of vapors below their respective threshold limit value. Ensure that eyewash station and safety shower is proximal to the work-station location.

Personal Protection

Splash goggles. Lab coat. Vapor respirator. Boots. Gloves. A MSHA/NIOSH approved respirator must be used to avoid inhalation of the product. Suggested protective clothing might not be sufficient; consult a specialist BEFORE handling this product.



Exposure Limits

This chemical is classified as a carcinogen. There is no acceptable exposure limit for a carcinogen.

Section IX. Physical and Chemical Properties				
Physical state @ 20°C	Liquid. (Clear, Colorless.)	Solubility	Slightly soluble in water (1.3g/100mL @ 20°C)	
Specific Gravity	1.33 (water=1)		Miscible in acetone, ether, alcohol, chloroform, carbon tetrachloride.	
Molecular Weight	84.93	Partition Coefficient	LOG P _{ow} 1.25	
Boiling Point	39°C (102.2°F)	Vapor Pressure	353.1 mmHg (@ 20°C)	
Melting Point	-95°C (-139°F)	Vapor Density	2.9 (Air = 1)	
Refractive Index	Not available.	Volatility	Not available.	
Critical Temperature	Not available.	Odor	Chloroform like.	
Viscosity	Not available.	Taste	Not available.	

Section X.	Stability and Reactivity Data
Stability	This material is stable if stored under proper conditions. (See Section VII for instructions)
Conditions of Instability	Avoid excessive heat and light.
Incompatibilities	Reactive with strong oxidizing agents, metals, strong alkalis (bases), aluminum, alkali metals, magnesium powders, potassium, sodium, concentrated nitric acid. This compound will attack some forms of plastic, rubber, and coatings.

Section XI. Toxicological Information

RTECS Number

PA8050000

Routes of Exposure

Eye Contact. Ingestion. Inhalation.

Toxicity Data

Rat LD₅₀ (oral) 1600 mg/kg Rabbit LD₅₀ (oral) 2000 mg/kg Rat LD₅₀ (inhalation) 52g/m³

Chronic Toxic Effects

CARCINOGENIC EFFECTS: Carcinogenic by RTECS criteria.

MUTAGENIC EFFECTS : Not available.

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Harmful if ingested or inhaled. Minimize exposure to this material. Severe overexposure can result in injury or death. Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening, or, occasionally, blistering.

Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

Section XII. Ecological Information

Ecotoxicity

Not available.

Environmental Fate

Dichloromethane's production and use as solvent, chemical intermediate, grain fumigant, paint stripper and remover, metal degreaser, and refrigerant may result in its release to the environment through various waste streams. If released to air, a vapor pressure of 435 mm Hg at 25 deg C indicates dichloromethane will exist solely as a vapor in the ambient atmosphere. Vapor-phase dichloromethane will be degraded in the atmosphere by reaction with photochemically-produced hydroxyl radicals; the half-life for this reaction in air is estimated to be 119 days. It will not be subject to direct photolysis. If released to soil, dichloromethane is expected to have very high mobility based upon an estimated Koc of 24. Volatilization from moist soil surfaces is expected to be an important fate process based upon a estimated Henry's Law constant of 3.25X10-3 atm-cu m/mole. Dichloromethane may volatilize from dry soil surfaces based upon its vapor pressure. Biodegradation in soil may occur based on activated sludge studies. If released into water, dichloromethane is not expected to adsorb to suspended solids and sediment in water based upon the estimated Koc. Biodegradation is possible in natural waters but will probably be very slow compared with evaporation. Volatilization from water surfaces is expected to be an important fate process based upon this compound's Henry's Law constant. Estimated volatilization half-lives for a model river and model lake are 1 hr and 4 days, respectively. An estimated BCF of 2 suggests the potential for bioconcentration in aquatic organisms is low. Hydrolysis is not an important degradation process under normal environmental conditions. Occupational exposure to dichloromethane may occur through inhalation and dermal contact with this compound at workplaces where dichloromethane is produced or used. The general population may be exposed to dichloromethane via inhalation of ambient air, ingestion of food and drinking water, and dermal contact with consumer products, such as paint strippers, which contain dichloromethane.

Section XIII. Disposal Considerations

Waste Disposal

Recycle to process, if possible. Consult your local regional authorities. You may be able to dissolve or mix material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber system. Observe all federal, state and local regulations when disposing of the substance.

Emergency phone number (800) 424-9300

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Section XIV.	Transport Information	
DOT Classification	DOT Class 6.1: Toxic material.	
PIN Number	UN1593	
Proper Shipping Name	Dichloromethane.	
Packing Group (PG)	III	
DOT Pictograms	POISON	

Section XV. Other Regulatory Information and Pictograms TSCA Chemical Inventory This compound is ON the EPA Toxic Substances Control Act (TSCA) inventory list. (EPA) WHMIS Classification CLASS D-1B: Material causing immediate and serious toxic effects (TOXIC). On DSL. (Canada) 200-838-9 EINECS Number (EEC) **EEC Risk Statements** R20/21/22- Harmful by inhalation, in contact with skin and if swallowed. R36/37/38- Irritating to eyes, respiratory system and skin. R45- May cause cancer. Japanese Regulatory Data ENCS No. (2)-36

Section XVI. Other Information

Version 1.0 Validated on 11/29/2007. Printed 11/29/2007.

Notice to Reader

TCI laboratory chemicals are for research purposes only and are NOT intended for use as drugs, food additives, households, or pesticides. The information herein is believed to be correct, but does not claim to be all inclusive and should be used only as a guide. Neither the above named supplier nor any of its subsidiaries assumes any liability whatsoever for the accuracy or completeness of the information contained herein. Final determination of suitability of any material is the sole responsibility of the user. All chemical reagents must be handled with the recognition that their chemical, physiological, toxicological, and hazardous properties have not been fully investigated or determined. All chemical reagents should be handled only by individuals who are familiar with their potential hazards and who have been fully trained in proper safety, laboratory, and chemical handling procedures. Although certain hazards are described herein, we can not guarantee that these are the only hazards which exist. Our MSDS sheets are based only on data available at the time of shipping and are subject to change without notice as new information is obtained. Avoid long storage periods since the product is subject to degradation with age and may become more dangerous or hazardous. It is the responsibility of the user to request updated MSDS sheets for products that are stored for extended periods. Disposal of unused product must be undertaken by qualified personnel who are knowledgeable in all applicable regulations and follow all pertinent safety precautions including the use of appropriate protective equipment (e.g. protective goggles, protective clothing, breathing equipment, facial mask, fume hood). For proper handling and disposal, always comply with federal, state, and local regulations.

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