



Material Safety Data Sheet

HAZARD WARNINGS	RISK PHRASES	PROTECTIVE CLOTHING
	Combustible material; avoid heat and sources of ignition. Harmful compound, minimize exposure. Irritating to skin, eyes, and the respiratory system. Lachrymator. This compound is a skin sensitizer. Risk of serious damage to eyes. Heat and light sensitive.	

Section I. Chemical Product and Company Identification				
Chemical Name	Acrylic Acid n-Butyl Ester (stabilized with MEHQ)			
Catalog Number	A0142	Supplier	TCI America 9211 N. Harborgate St.	
Synonym	2-Propenoic Acid Butyl Ester		Portland OR 1-800-423-8616	
Chemical Formula	CH ₂ :CHCOOC ₄ H ₉			
CAS Number	141-32-2	In case of Emergency Call	Chemtrec® (800) 424-9300 (U.S.) (703) 527-3887 (International)	

Section II. Composition and Information on Ingredients					
Chemical	Name	CAS Number	Percent (%)	TLV/PEL	Toxicology Data
Acrylic Acid n-Butyl Ester (stabilized with MEHQ)		141-32-2	Min. 99.0 (GC)		Rat LD ₅₀ (oral) 900mg/kg Rat LC ₅₀ (inhalation) 2730ppm/4H Rabbit LD ₅₀ (dermal) 2ml/kg

Acute Health Effects Harmful if ingested or inhaled. Minimize exposure to this material. Severe overexposure can result in injury or death.

Irritating to eyes and skin on contact. Inhalation causes irritation of the lungs and respiratory system. Inflammation of the eye is characterized by redness, watering, and itching. Skin inflammation is characterized by itching, scaling, reddening,

or, occasionally, blistering.

Skin contact may result in sensitization. Always cover all exposed skin with an impermeable layer and use proper eye protection. A OSHA/MSHA approved dust and vapor respirator is required when working with this material. Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

Chronic Health Effects CARCINOGENIC EFFECTS: Not available.

MUTAGENIC EFFECTS: Not available.

TERATOGENIC EFFECTS: Not available.

DEVELOPMENTAL TOXICITY: REPRODUCTIVE EFFECTS:

Rat TCLo (inhalation) 135ppm/6 Hours, female, 6-15 Days of pregnancy.

Toxic Effects:

Effects on Fertility- Post-implantation mortality.

Rat TCLo (inhalation) 200ppm/6 Hours, female, 6-20 Days of pregnancy.

Toxic Effects:

Effects on Embryo or Fetus- Fetotoxicity.

Rat TCLo (inhalation) 200ppm, female, 6-20 Days of pregnancy.

Toxic Effects:

Effects on Embryo or Fetus- Fetotoxicity.

Maternal Effects- Other effects.

Repeated or prolonged exposure to this compound is not known to aggravate existing medical conditions.

Section IV.	First Aid Measures				
Eye Contact	Check for and remove any contact lenses. In case of contact, immediately flush eyes with plenty of water for at least 15 minutes. Get medical attention.				
Skin Contact	In case of contact, immediately flush skin with plenty of water. Remove contaminated clothing and shoes. Wash clothing before reuse. Thoroughly clean shoes before reuse. Get medical attention.				
Inhalation	If the victim is not breathing, perform mouth-to-mouth resuscitation. Loosen tight clothing such as a collar, tie, belt or waistband. If breathing is difficult, oxygen can be administered. Seek medical attention if respiration problems do not improve.				
Ingestion	INDUCE VOMITING by sticking finger in throat. Lower the head so that the vomit will not reenter the mouth and throat. Loosen tight clothing such as a collar, tie, belt or waistband. If the victim is not breathing, perform mouth-to-mouth resuscitation. Examine the lips and mouth to ascertain whether the tissues are damaged, a possible indication that the toxic material was ingested; the absence of such signs, however, is not conclusive.				

Section IX. Physical and Chemical Properties						
Physical state @ 20°C	Liquid. (Clear, colorless.)	Solubility	Solubility in water @ 20°C: 0.14g/100ml; @ 40°C: 0.12g/100ml.			
Specific Gravity	0.90 (water=1)	_	Solubility of water in n-Butyl acrylate @ 20°C: 0.8ml/100g.			
Molecular Weight	128.17	Partition Coefficient	Log K _{ow} = 2.36			
Boiling Point	145 to 146°C (293 to 294.8°F)	Vapor Pressure	0.4 kPa (@ 20°C)			
Melting Point	Freezing Point: -64°C (-83.2°F)	Vapor Density	4.42 (Air = 1)			
Refractive Index	1.4185	Volatility	Not available.			
Critical Temperature	Not available.	Odor	Strong, fruity odor.			
Viscosity	0.75Pas @ 20°C.	Taste	Not available.			

Section X. Stability and Reactivity Data

This material is stable if stored under proper conditions. (See Section VII for instructions)

Conditions of Instability Avoid excessive heat and light.

Incompatibilities Reactive with strong oxidizing agents, strong acids, strong alkalis (bases).

The product may undergo dangerous decomposition, condensation or polymerization, it may react violently with water to emit toxic gases or it may become self-reactive under conditions of shock or increase in temperature or pressure.

Stability

(stabilized with MEHQ)

Section XI. Toxicological Information

RTECS Number UD3150000

Routes of Exposure Eye Contact. Ingestion. Inhalation.

Toxicity Data Rat LD₅₀ (oral) 900mg/kg

 $\begin{array}{l} Rat\ LD_{50}\ \ (oral)\ 900mg/kg \\ Rat\ LC_{50}\ \ (inhalation)\ 2730ppm/4H \\ Rabbit\ LD_{50}\ \ (dermal)\ 2ml/kg \end{array}$

Chronic Toxic Effects CARCINOGENIC EFFECTS: Not available.

MUTAGENIC EFFECTS: Not available.
TERATOGENIC EFFECTS: Not available.

DEVELOPMENTAL TOXICITY: REPRODUCTIVE EFFECTS:

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or, occasionally, blistering.

Skin contact may result in sensitization. Always cover all exposed skin with an impermeable layer and use proper eye protection. A OSHA/MSHA approved dust and vapor respirator is required when working with this material.

Follow safe industrial hygiene practices and always wear proper protective equipment when handling this compound.

Section XII. Ecological Information

Ecotoxicity Not available.

Environmental Fate

Butyl acrylate's production and use as a polymer and copolymer for solvent coatings, adhesives, paints, binders and emulsifiers may result in its release to the environment through various waste streams. If released to air, a vapor pressure of 5.45 mm Hg at 25 deg C indicates butyl acrylate will exist solely as a vapor in the ambient atmosphere. Vapor-phase butyl acrylate will be degraded in the atmosphere by reaction with photochemically-produced hydroxyl radicals; the half-life for this reaction in air is estimated to be 1.2 days. Because the structurally similar compound ethyl acrylate does not absorb light in the environmental UV spectrum (>290 nm), butyl acrylate is not expected to directly photolyze. If released to soil, butyl acrylate is expected to have high mobility based upon a Koc of 88. Volatilization from moist soil surfaces is expected to be an important fate process based upon an estimated Henry's Law constant of 6.6X10-4 atm-cu m/mole. Butyl acrylate may volatilize from dry soil surfaces based upon its vapor. If released into water, butyl acrylate is not expected to adsorb to suspended solids and sediment in the water column based upon the estimated Koc. Butyl acrylate may be susceptible to biodegradation; in screening tests it reached 61% of its theoretical BOD in 2 weeks using an activated sludge inoculum. Volatilization from water surfaces is expected to be an important fate process based upon this compound's estimated Henry's Law constant. Estimated volatilization half-lives for a model river and model lake are 5 hours and 5 days, respectively. An estimated BCF of 37 suggests the potential for bioconcentration in aquatic organisms is moderate. Hydrolysis of butyl acrylate may be a significant process based upon a hydrolytic half-life of 4 hours at pH 11; half-lives at pH 7,8 and 9 were 4 years, 150 days and 15 days, respectively. Occupational exposure to butyl acrylate may occur through inhalation and/or dermal contact with this compound at workplaces where butyl acrylate is produced or used. The general population may be exposed to butyl acrylate via inhalation or dermal contact with furniture surface coatings, as well as inhalation of ambient air near butyl acrylate manufacturing facilities. (HSDB)

Section XIII. Disposal Considerations

Waste Disposal

Recycle to process, if possible. Consult your local regional authorities. You may be able to dissolve or mix material with a combustible solvent and burn in a chemical incinerator equipped with an afterburner and scrubber system. Observe all federal, state and local regulations when disposing of the substance.

Section XIV. Transport Information

DOT Classification CLASS 3: Flammable liquid.

PIN Number UN2348

Proper Shipping Name Butyl acrylates, stabilized

Packing Group (PG)

DOT Pictograms



Emergency phone number (800) 424-9300

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(stabilized with MEHQ)

Section XV. Other Regulatory Information and Pictograms

TSCA Chemical Inventory This compound is **ON** the EPA Toxic Substances Control Act (TSCA) inventory list.

(EPA) This product is subject to SARA Section 313 reporting requirements.

WHMIS Classification CLASS B-3: Combustible liquid with a flash point between 37.8°C (100°F) and 93.3°C (200°F).

(Canada) CLASS D-1A: Material causing immediate and serious toxic effects (VERY TOXIC).

CLASS D-2B: Material causing other toxic effects (TOXIC).

CLASS F: Dangerously reactive material.

On DSL.

EINECS Number (EEC) 205-480-7

EEC Risk Statements R10- Flammable.

R20/21/22- Harmful by inhalation, in contact with skin and if swallowed.

R36/37/38- Irritating to eyes, respiratory system and skin.

R41- Risk of serious damage to eyes.

R43- May cause sensitization by skin contact.

Japanese Regulatory Data ENCS No. 2-989, 6-2010

Section XVI. Other Information

Version 1.0 Validated on 9/13/2004. Printed 1/10/2005.

Notice to Reader

TCI laboratory chemicals are for research purposes only and are NOT intended for use as drugs, food additives, households, or pesticides. The information herein is believed to be correct, but does not claim to be all inclusive and should be used only as a guide. Neither the above named supplier nor any of its subsidiaries assumes any liability whatsoever for the accuracy or completeness of the information contained herein. Final determination of suitability of any material is the sole responsibility of the user. All chemical reagents must be handled with the recognition that their chemical, physiological, toxicological, and hazardous properties have not been fully investigated or determined. All chemical reagents should be handled only by individuals who are familiar with their potential hazards and who have been fully trained in proper safety, laboratory, and chemical handling procedures. Although certain hazards are described herein, we can not guarantee that these are the only hazards which exist. Our MSDS sheets are based only on data available at the time of shipping and are subject to change without notice as new information is obtained. Avoid long storage periods since the product is subject to degradation with age and may become more dangerous or hazardous. It is the responsibility of the user to request updated MSDS sheets for products that are stored for extended periods. Disposal of unused product must be undertaken by qualified personnel who are knowledgeable in all applicable regulations and follow all pertinent safety precautions including the use of appropriate protective equipment (e.g. protective clothing, breathing equipment, facial mask, fume hood). For proper handling and disposal, always comply with federal, state, and local regulations.

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